



Spontaneously blinking spiroamide rhodamines for live SMLM imaging of the plasma membrane†

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We have developed spontaneously blinking fluorescent probes based on the reversible spirolactamization of rhodamine, to efficiently image the plasma membrane (PM) of live cells with enhanced resolution using SMLM. This study demonstrates that the blinking efficiency of spiroamide PM probes is not solely governed by their pK_a ; the presence of a charged polar group on the amide should also be taken into account.

The advent of super-resolution imaging techniques based on single-molecule localization microscopy (SMLM) has revolutionized optical microscopy as it enables the visualization of biological structures with unprecedented resolution.^{1,2} While several organelles and lipid-based cellular structures have been successfully imaged in both fixed and live samples,^{3,4} SMLM imaging of the plasma membrane (PM) remains highly constrained.^{5,6} Specifically, live-cell SMLM imaging of the PM poses significant challenges due to the free diffusion of fluorescent probes within the lipid bilayer and the inherent movement of the PM during image acquisition. As a result, examples of live SMLM imaging of the PM are exceedingly rare and largely restricted to PAINT-based approaches that utilize transient binding of moderately lipophilic dyes.^{7–9} This underscores the pressing need for efficient spontaneously blinking probes to advance live SMLM imaging of the PM. In response to this challenge, we have directed our recent research efforts toward developing spontaneously blinking probes tailored for this purpose. Notably, we achieved 3D live SMLM of the PM using directed photooxidation of photoconvertible PM probes.¹⁰ Recently, we also introduced a novel concept involving a self-triggered photooxidation cascade of leuco-rhodamine derivatives.¹¹ Indeed, live SMLM imaging demands probes with a spontaneous blinking mechanism

characterized by a bright “on” state, a dark “off” state, and a low density of emissive molecules at any given time.¹² In this context, rhodamines featuring an intramolecular nucleophilic moiety are particularly well-suited due to their dynamic equilibrium between an emissive open form and a non-emissive spirocyclic form.¹³ The Urano group has made significant contributions to this field, leveraging hydroxyl groups as intramolecular nucleophiles in various rhodamine derivatives, including Si-,¹⁴ C-,¹⁵ and O-rhodamines,¹⁶ offering a wide range of spectral properties and finely tuned spirocyclization equilibria. In particular, spiroamide O-rhodamines have been extensively studied and shown good efficacies in nanoscopy applications.^{17–19} Importantly, several studies report that special attention must be given to the design of these probes to reach the most appropriate blinking properties depending on the cellular target.²⁰ Notably, since spiroamide opening depends on the pH, the pK_a of the probes is the main factor that governs this fluorogenic equilibrium.^{21,22}

In this work, we investigated the performance of three potential spontaneously blinking spiroamide rhodamine (SR) probes with increasing pK_a values and evaluated their efficacy for rapid imaging of live cell contours using SMLM (Fig. 1A). To this end, we first synthesized a clickable rhodamine derivative, R-COOH (see ESI†), which was coupled to three different amines to produce the spiroamide rhodamine SR probes (Fig. 1B). The choice of amines was informed by previous studies reporting SR probes based on rhodamine 6G with pK_a values of $<4^{23}$ and 2.8^{24} when coupled to ethylenediamine and ethanolamine, respectively. In contrast, 1-adamantylamine, owing to its bulkiness, yielded a pK_a of 6.5^{24} . Once synthesized, the spiroamide rhodamines were converted into PM probes (SR-PM-NH₂, SR-PM-OH, and SR-PM-Ad) by coupling them with CAZ, a clickable amphiphilic zwitterion. CAZ, when used in pairs, mimics phospholipids and intercalates into the lipid bilayer, acting as an efficient plasma membrane targeting moiety.^{6,25,26}

First, the photophysical properties of the SR precursors were determined and are presented in Fig. 2 (table). As expected, under acidic conditions (pH 2), the probes exhibited absorption with a λ_{Abs} max around 560 nm and emission with a λ_{Em} max

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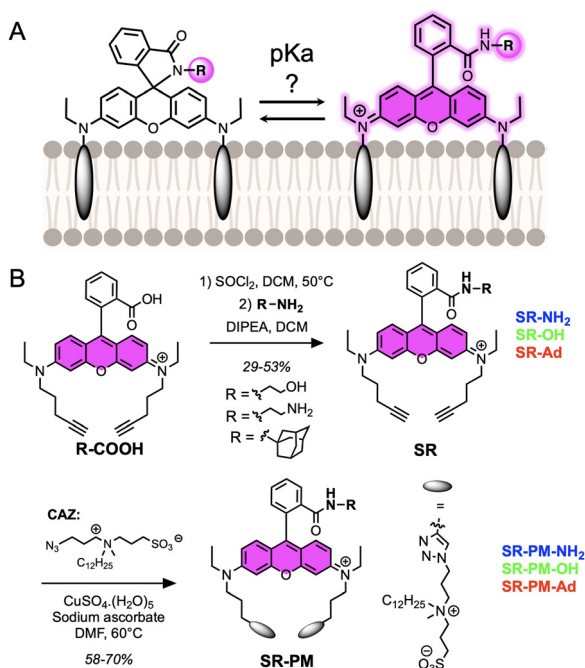


Fig. 1 (A) Principle and (B) synthesis of the blinking SR-PM probes.

around 582 nm, along with similar fluorescence quantum yields. This indicates that the coupled amine does not interfere with the fluorescence process. At a pH close to physiological extracellular conditions (pH 7), their absorbance and emission intensities in the visible region decreased significantly due to spirolactamization. The fluorescence spectra of the three SR probes were recorded at various pH values (Fig. 2A), providing

SR Probe	pH	$\lambda_{\text{max, Abs}}$ (nm)	$\lambda_{\text{max, Em}}$ (nm)	ϵ (M ^{−1} ·cm ^{−1}) ^a	ϕ_F ^b	pK _a
SR-NH ₂	2	560	583	33 480	0.46	2.06
	7	N.A. ^a	N.A. ^a	N.A. ^c	N.A. ^c	
SR-OH	2	561	582	29 710	0.50	2.82
	7	N.A. ^a	N.A. ^a	N.A. ^c	N.A. ^c	
SR-Ad	2	563	584	95 480	0.45	5.09
	7	560	583	3 840	0.40	

^a Apparent extinction coefficient, ^b Does not absorb nor emit in the visible, ^c Fluorescence quantum yield

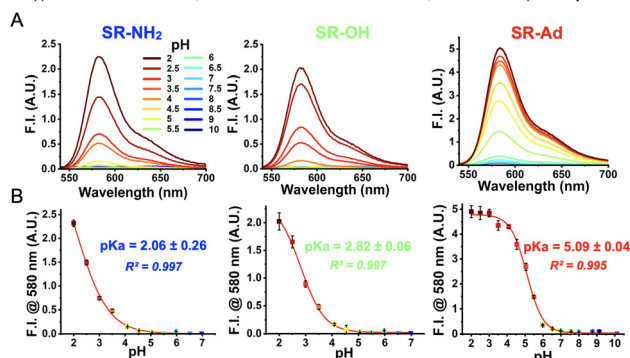


Fig. 2 (table) Photophysical properties of SR probes in controlled pH buffer/ethanol (7 : 3) at a concentration of 1 μM . (A) Emission spectra of the SR probes at various pH levels with an excitation wavelength of 530 nm. (B) pH titration curves of the SR probes, fitted with a sigmoidal dose–response equation to determine their pK_a values.

their respective pK_a values (Fig. 2B). While SR-Ad showed a lower pK_a compared to previously reported SR probes based on rhodamine 6G (5.09 versus 6.4),²⁴ SR-NH₂ and SR-OH displayed significantly lower pK_a values (2.06 and 2.82, respectively), consistent with their rhodamine 6G analogues.^{23,24} Before proceeding with live cellular imaging, a viability test was performed on HeLa cells, revealing that the three SR-PM probes did not induce any observable cytotoxicity at 200 nM (Fig. S1, ESI†).

The probes were first evaluated using laser scanning confocal microscopy (Fig. S2, ESI†). As expected, owing to their low pK_a values favoring the off-spiroform, the resulting images displayed weak signal-to-noise ratios along with low Pearson's colocalization coefficients compared with a reference PM probe (MemBright Cy5.5⁶). Interestingly, while cells in the presence of SR-PM-Ad and SR-PM-OH exhibited a dim fluorescence signal mainly due to autofluorescence, those stained with SR-PM-NH₂ showed a weak additional signal at the cell surface and the highest Pearson's colocalization coefficient, despite its lowest pK_a value among the three probes. This observation suggests that factors beyond pK_a influence the ring-opening equilibrium at the PM.

Single-molecule localization microscopy (SMLM) imaging was then performed on live HeLa cells in the presence of 200 nM SR-PM probes (Fig. 3 and Supplementary Movies, ESI†) to rapidly image their contours with enhanced resolution. The results revealed significant differences. As expected, SR-PM-OH, due to its low pK_a, produced virtually no blinking events and failed to reconstruct the cell surface in SMLM (Fig. 3A). Surprisingly, SR-PM-Ad, despite being \approx 180-fold more basic, provided only a slightly higher number of blinking events per cell surface (17, Fig. 3B). Once again, SR-PM-NH₂ distinguished itself with unique behavior, generating significantly more blinking events

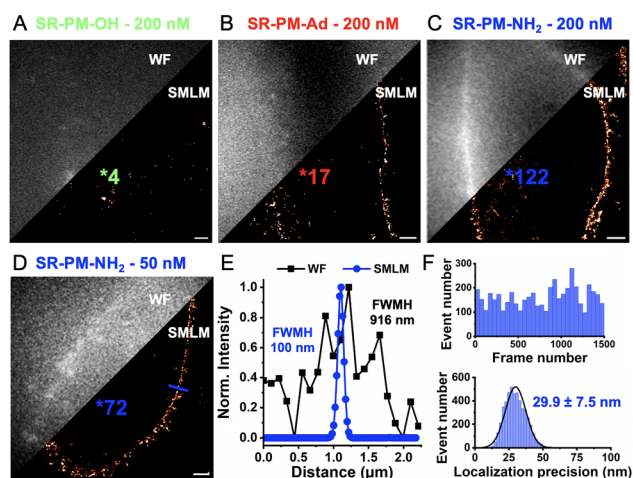


Fig. 3 Live SMLM. (A)–(C) Widefield and corresponding reconstructed SMLM images of HeLa cells in the presence of SR-PM probes (200 nM) and obtained after 14 s acquisition (1000 frames at 13.9 ms exposure time). (D) Widefield and SMLM images of a HeLa cell using SR-PM-NH₂ at 50 nM. The colored number on the SMLM image is the average number of blinking events per frame per PM surface (1500 frames at 13.9 ms exposure time). (E) Plot profile corresponding to the blue line in D. (F) Single-molecule statistics obtained with SR-PM-NH₂ (50 nM) in the HeLa cell membrane. Scale bar is 5 μm .

in the single-molecule regime and successfully reconstructing the cell surface of live HeLa cells within only 14 seconds of acquisition (Fig. 3C).

Although the bulk concentration of the probe might seem low (200 nM), the local concentration, due to accumulation in the membrane, can become relatively high. Consequently, the concentration of SR-PM-NH₂ was reduced to 50 nM. At this reduced concentration, the number of events per cell surface decreased; however, the resulting SMLM images improved, likely due to reduced phototoxicity and lower blinking density, and a clear resolution enhancement was achieved compared to widefield microscopy (Fig. 3E). This improvement was observed in only 21 seconds of acquisition (1500 frames, 13.9 ms per frame) (Fig. 3F). At a 532 nm laser irradiance of 500 W cm⁻², SR-PM-NH₂ provided a stable number of blinks per frame with a localization precision of 29.9 ± 7.5 nm (Fig. 3F).

To understand the enhanced performance of SR-PM-NH₂ compared to the other SR-PM probes, additional experiments were conducted. Initially, we aimed to assess their opening and closing kinetics. While the ring-closure kinetics could not be determined due to the rapidity of the process (Fig. S3, ESI†), the ring-opening kinetics could be measured (Fig. 4A). As expected, based on its pK_a value, SR-Ad exhibited the highest ring-opening rate from pH 7.4 to 2 of 0.031 s⁻¹. Conversely, SR-OH and SR-NH₂, which have comparable low pK_a values, displayed much lower ring-opening kinetics of ≈ 3–4 × 10⁻⁴ s⁻¹. These results were expected and did not provide further insights into the differences observed in SMLM performance between SR-PM-OH and SR-PM-NH₂. To gain valuable insights into the differences in probe performance, we attempted to investigate the influence of environmental polarity on the spirocyclization equilibrium using water/dioxane mixtures, as previously reported.²⁷

However, due to their pronounced lipophilic nature, the SR probes exhibited poor solubility under these conditions, preventing such an analysis. To address these limitations, the reaction described in Fig. 1A can be decomposed into a three-step process based on (1) a ring-opening process, (2) a protonation reaction and (3) a rotation of the amide group (see Scheme S1, ESI†). The ring-opening process being expected to be the rate-determining step,¹⁹ the Gibbs free energies of the former reaction (Fig. 4B) were calculated with DFT using an implicit solvent computational model to qualitatively compare the activation energies of the two compounds.²⁸ For this reaction step, it is worth noting that the relative pK_a values of the two probes have no impact on these results. The calculated barriers revealed significant differences, indicating that SR-NH₂ requires considerably less energy to open compared to SR-OH. This is likely due to the presence of the ammonium group in the protonated form of SR-NH₂ (Fig. 4E), which stabilizes the highly polar open-ring zwitterionic structure and lowers its relative free energy.

Finally, we hypothesized that the ring-opening efficiency might be influenced by the membrane environment. Indeed, in this interfacial environment, factors other than pH could govern the ring opening of SR. To explore this, 80 nm large unilamellar vesicles (LUVs) composed of neutral zwitterionic DOPC and anionic DOPS were prepared at various ratios: 1/0, 1/1, and 0/1 (Fig. S4, ESI†). The probes were incubated with the LUVs at a fixed pH (PBS, pH 7.4), and their absorption and emission spectra were recorded (Fig. 4C and D). The results revealed that within these membrane models, the pK_a was not the primary determinant of ring opening, as the intensity of the absorption signal at 560 nm—indicative of ring opening—did not correlate with the pK_a values. For instance, SR-PM-Ad, which has the highest pK_a (5.09), predominantly existed in its closed form (Fig. S5, ESI†). Interestingly, when embedded in the

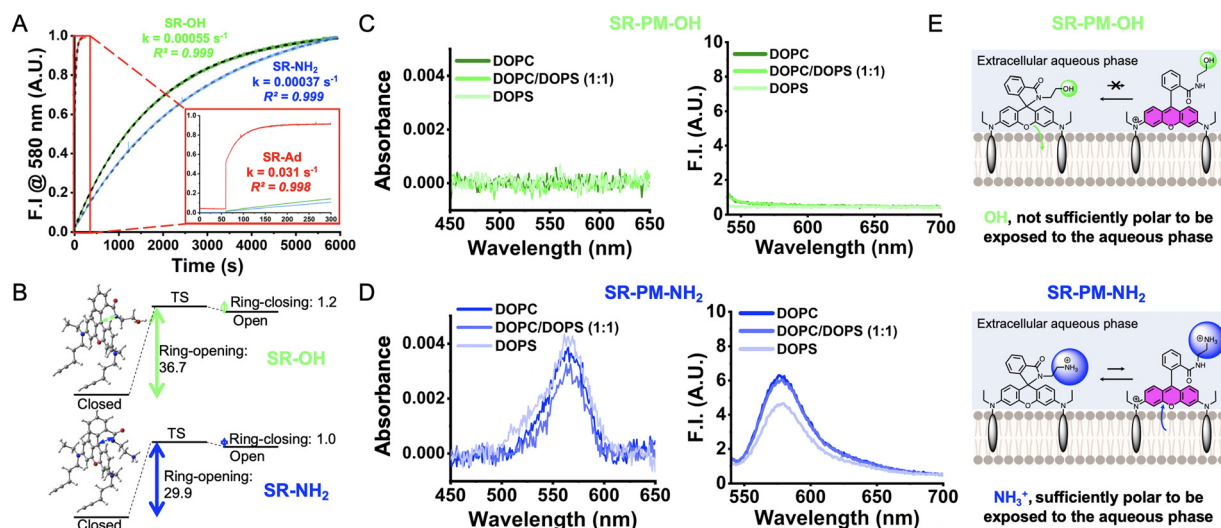


Fig. 4 Other parameters that govern the opening of SR probes. (A) Monitoring of the fluorescence intensity @ λ_{Em} max of the SR probes over time after acidification (from pH 7.4 to 2) providing the kinetic rates of ring opening. (B) DFT calculation providing the Gibbs free energies (in kcal mol⁻¹) required for the ring-opening and ring-closing of SR-OH and SR-NH₂. (C) and (D) Absorption (left) and emission (right) spectra of SR-PM probes when incubated with LUVs of various compositions. The probe lipid ratio was set to 1:200. (E) Schematic representation of the phenomenon driving the ring opening of SR-PM at the lipid–water interface.



negatively charged DOPS bilayer, a signal appeared, suggesting that the negative charges of the lipid polar heads can stabilize the cationic rhodamine after ring opening (Fig. S5, ESI†). Notably, SR-PM-NH₂, despite having the lowest pK_a, exhibited the highest absorbance (Fig. 4C) and fluorescence intensity (Fig. 4D) in LUVs, irrespective of their composition. In contrast, SR-PM-OH, which has a higher pK_a, showed neither absorbance nor fluorescence signals in any type of LUV (Fig. 4C). These results suggest that at the lipid bilayer interface, SR probes with non-polar amides, even those with high pK_a values (SR-PM-Ad), tend to orient within the lipid phase, favoring their hydrophobic closed form (Fig. 4E). SR-PM-NH₂, on the other hand, contains a polar ammonium (NH₃⁺) at physiological pH, which promotes the orientation of the rhodamine toward the aqueous phase (Fig. 4E). This hypothesis was confirmed by monitoring the fluorescence intensity at 580 nm upon the addition of SR-PM to a solution of DOPC and DOPS LUVs (Fig. S6, ESI†). While SR-PM-OH showed no signal over time, SR-PM-NH₂ produced an immediate and constant low signal. Interestingly, SR-PM-Ad rapidly reached its maximum intensity before decreasing with a kinetics depending on the LUV composition, indicating that its open form is stabilized by the negatively charged phosphatidylserine before ring closure (Fig. S6, ESI†). Overall, at the lipid-water interface, SR probes with non-charged and non-polar amides (SR-OH and SR-Ad) experience a shift in their apparent pK_a to much lower values. In contrast, SR-PM-NH₂, which primarily orients toward the extracellular aqueous phase, retains its ability to protonate, thereby enabling a higher number of transient and reversible ring openings at the cell surface (Fig. 4E). In conclusion, we have demonstrated that SR-PM-NH₂ is an efficient spontaneously blinking fluorescent probe for imaging the plasma membrane and contours of live cells with short acquisition times (<30 s) and enhanced resolution using SMLM. Its superior performance can be attributed not only to its low pK_a, which ensures the low density of emissive dye required for SMLM, but also to the presence of a charged ammonium group, which: (1) lowers the free energy barrier for ring opening, and (2) promotes the orientation of the rhodamine moiety toward the aqueous phase, where ring opening is favored compared to being embedded in lipids. This study underscores the importance of considering structural and physical properties beyond pK_a when designing spontaneously blinking fluorescent probes based on spirolactamization for SMLM, tailored to their intended cellular localization.

Conceptualization SP, MC; formal analysis SP, SW, AP; funding acquisition MC; investigation SP, SW; project administration MC; supervision MC; validation MC; visualization SP MC; writing SP, AP, MC.

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Data availability

The data supporting this article have been included as part of the ESI†.

Conflicts of interest

There are no conflicts to declare.

Notes and references

- 1 M. Lelek, M. T. Gyparaki, G. Beliu, F. Schueder, J. Griffié, S. Manley, R. Jungmann, M. Sauer, M. Lakadamyali and C. Zimmer, *Nat. Rev. Methods Primers*, 2021, **1**, 39.
- 2 T. Klein, S. Proppert and M. Sauer, *Histochem. Cell Biol.*, 2014, **141**, 561–575.
- 3 X. Chen, X. Wang, F. Huang and D. Ma, *Photonix*, 2024, **5**, 29.
- 4 A. Jimenez, K. Friedl and C. Leterrier, *Methods*, 2020, **174**, 100–114.
- 5 M. Collot, S. Pfister and A. S. Klymchenko, *Curr. Opin. Chem. Biol.*, 2022, **69**, 102161.
- 6 M. Collot, P. Ashokkumar, H. Anton, E. Boutant, O. Faklaris, T. Galli, Y. Mély, L. Danglot and A. S. Klymchenko, *Cell Chem. Biol.*, 2019, **26**, 600–614.e7.
- 7 D. I. Danylchuk, S. Moon, K. Xu and A. S. Klymchenko, *Angew. Chem., Int. Ed.*, 2019, **58**, 14920–14924.
- 8 I. O. Aparin, R. Yan, R. Pelletier, A. A. Choi, D. I. Danylchuk, K. Xu and A. S. Klymchenko, *J. Am. Chem. Soc.*, 2022, **144**, 18043–18053.
- 9 R. Yan, K. Chen and K. Xu, *J. Am. Chem. Soc.*, 2020, **142**, 18866–18873.
- 10 L. Saladin, V. Breton, V. Le Berruyer, P. Nazac, T. Lequeu, P. Didier, L. Danglot and M. Collot, *J. Am. Chem. Soc.*, 2024, **146**, 17456–17473.
- 11 S. Pfister, V. Le Berruyer, K. Fam and M. Collot, *bioRxiv*, 2024, 2024.05.28.596159.
- 12 H. Li and J. C. Vaughan, *Chem. Rev.*, 2018, **118**, 9412–9454.
- 13 D. Si, Q. Li, Y. Bao, J. Zhang and L. Wang, *Angew. Chem., Int. Ed.*, 2023, **62**, e202307641.
- 14 S. Uno, M. Kamiya, T. Yoshihara, K. Sugawara, K. Okabe, M. C. Tarhan, H. Fujita, T. Funatsu, Y. Okada, S. Tobita and Y. Urano, *Nat. Chem.*, 2014, **6**, 681–689.
- 15 R. Tachibana, M. Kamiya, A. Morozumi, Y. Miyazaki, H. Fujioka, A. Nanjo, R. Kojima, T. Komatsu, T. Ueno, K. Hanaoka, T. Yoshihara, S. Tobita and Y. Urano, *Chem. Commun.*, 2020, **56**, 13173–13176.
- 16 S. Uno, M. Kamiya, A. Morozumi and Y. Urano, *Chem. Commun.*, 2018, **54**, 102–105.
- 17 V. N. Belov, M. L. Bossi, J. Fölling, V. P. Boyarskiy and S. W. Hell, *Chem. – Eur. J.*, 2009, **15**, 10762–10776.
- 18 P. J. Macdonald, S. Gayda, R. A. Haack, Q. Ruan, R. J. Himmelsbach and S. Y. Tetin, *Anal. Chem.*, 2018, **90**, 9165–9173.
- 19 Q. Qi, W. Chi, Y. Li, Q. Qiao, J. Chen, L. Miao, Y. Zhang, J. Li, W. Ji, T. Xu, X. Liu, J. Yoon and Z. Xu, *Chem. Sci.*, 2019, **10**, 4914–4922.
- 20 Z. Ye, H. Yu, W. Yang, Y. Zheng, N. Li, H. Bian, Z. Wang, Q. Liu, Y. Song, M. Zhang and Y. Xiao, *J. Am. Chem. Soc.*, 2019, **141**, 6527–6536.
- 21 Q. Qiao, W. Liu, J. Chen, X. Wu, F. Deng, X. Fang, N. Xu, W. Zhou, S. Wu, W. Yin, X. Liu and Z. Xu, *Angew. Chem., Int. Ed.*, 2022, **61**, e202202961.
- 22 Z. Ye, Y. Zheng, X. Peng and Y. Xiao, *Anal. Chem.*, 2022, **94**, 7990–7995.
- 23 Z. Li, S. Wu, J. Han and S. Han, *Analyst*, 2011, **136**, 3698–3706.
- 24 L. Yuan, W. Lin and Y. Feng, *Org. Biomol. Chem.*, 2011, **9**, 1723–1726.
- 25 M. Collot, E. Boutant, M. Lehmann and A. S. Klymchenko, *Bioconjugate Chem.*, 2019, **30**, 192–199.
- 26 S. Pfister, J. Lesieur, P. Bourdoncle, M. Elhassan, P. Didier, N. Anton, H. Anton and M. Collot, *Anal. Chem.*, 2024, **96**, 12784–12793.
- 27 N. Lardon, L. Wang, A. Tschanz, P. Hoess, M. Tran, E. D'Este, J. Ries and K. Johnsson, *J. Am. Chem. Soc.*, 2021, **143**, 14592–14600.
- 28 R. Tachibana, M. Kamiya, A. Morozumi, Y. Miyazaki, H. Fujioka, A. Nanjo, R. Kojima, T. Komatsu, T. Ueno, K. Hanaoka, T. Yoshihara, S. Tobita and Y. Urano, *Chem. Commun.*, 2020, **56**, 13173–13176.

